Tribhuvan University Institute of Science and Technology
Computer Science and Information Technology (B.Sc.CSIT)
Model Question for Entrance Examination

Full Marks: 100
Attempt all questions. Time: 2 hrs.

## Mathematics

$(25 \times 1=25)$

1. Number of nonempty singleton subsets from the elements of the set $\{1$, $\mathrm{m}, \mathrm{n}, \mathrm{p}\}$ is
(a) 4
(b) 3
(c) 2
(d) 1
2. Let $f: \mathrm{A} \rightarrow \mathrm{B}$, then $f$ is invertible if
(a) $f$ is one-one
(b) $f$ is onto
(c) $f$ is both one-one and onto
(d) $f$ is one-one into
3. Let $\mathrm{f}: \mathrm{R} \rightarrow \mathrm{R}$ be defined by $f(x)=3 x-4$, then $f^{-1}(x)$ is
(a) $\frac{x+4}{3}$
(b) $\frac{x-4}{3}$
(c) $x+4$
(d) $x-4$
4. $\lim _{x \rightarrow \infty}\left(1+\frac{3}{x}\right)^{x}$ is equal to
(a) $e^{3}$
(b) 3 e
(c) $3 e^{3}$
(d) $1 / e^{3}$
5. $\lim _{x \rightarrow \infty} \frac{\sin x}{x}$ is equal to
(a) 3
(b) 2
(c) 1
(d) 0
6. If the function $f(x)=\left\{\begin{array}{cl}2 x, & x<2 \\ 2, & x=2 \\ x^{2}, & x>2\end{array}\right.$, then the function has
(a) Continuous at $x=2$
(b) discontinuous at $x=2$
(c) removable discontinuous at $x=2$
(d) does not exist

B.Sc. CSIT Entrance Examination Model Question
7. $\frac{d}{d x}(\log |x|)$ is equal to
(a) $\frac{1}{|x|}$
(b) $\pm \frac{1}{x}$
(c) $\frac{1}{x}$
(d) $\frac{1}{x^{2}}$
8. Derivative of $\sin ^{3} x$ w.r.t. $\cos ^{3} x$ is
(a) $\tan x$
(b) $-\tan x$
(c) $\tan ^{3} x$
(d) $\cot x$
9. If $f(\mathrm{x})=\log (\log \mathrm{x})$, then $f^{1}(\mathrm{e})$ is equal to
(a) e
(b) $1 / \mathrm{e}$
(c) 1
(d) -1
10. The minimum value of $f(x)=\sin x \cdot \cos x$ is
(a) $-1 / 2$ (b) $1 / 2$
(c) 0
(d) 1
11. $\int \log x d x$ is equal to
(a) $(\log x-1)+C$
(b) $(x \log x-1)+C$
(c) $(x \log x+1)+C$
(d) $x(\log x-1)+C$
12. $\int_{0}^{\sqrt{3} / 2} \frac{d x}{\sqrt{1}-x^{2}}$ is equal to
(a) $\pi / 3$
(b) $\pi / 6$
(c) $\pi / 4$
(d) $\pi / 2$
13. The area bounded by the parabola $y^{2}=x$, the line $y=4$ and $y$-axis is
(a) $16 / 3$
(b) $32 / 3$
(c) $64 / 3$
(d) $128 / 3$
14. The modulus of $\frac{(1-i)^{3}}{1-i^{3}}$ is equal to
(a) -1
(b)i
(c) 0
(d) -2
15. The roots of the equation $2 x^{2}-3 x+1=0$ are
(a) real, unequal, rational
(b) real, equal
(b) perfect square, equal
(d) unequal, imaginary
16. If $A=\left(\begin{array}{ll}1 & 2 \\ 3 & 4\end{array}\right)$ and $B=\left(\begin{array}{ll}5 & 6 \\ 7 & 8\end{array}\right)$, then $(\mathrm{AB})^{\prime}$ is
(a) $\left(\begin{array}{ll}1 & 0 \\ 0 & 1\end{array}\right)$
(b) $\left(\begin{array}{ll}19 & 43 \\ 22 & 50\end{array}\right)$
(c) $\left(\begin{array}{ll}19 & 22 \\ 43 & 50\end{array}\right)$
(d) $\left(\begin{array}{ll}22 & 50 \\ 43 & 19\end{array}\right)$
17. The two lines of the system $6 x-4 y=10,3 x-2 y=5$ are
(a) parallel
(b) intersecting
(c) coincident
(d) independent
18. If the A.M. and G.M between two given numbers are 81 and 18 , respectively, then H.M. is
(a) 99
(b) 49.5
(c) 4.5
(d) 4
19. The most general value of $\theta$ satisfying the equation $3 \tan ^{2} \theta=1$ is
(a) $\mathrm{n} \pi+\pi / 6$
(b) $n \pi \pm \pi / 6$
(c) $2 \mathrm{n} \pi \pm \pi / 6$
(d) $(-1)^{3} \alpha+\pi / 6$
20. $\tan ^{-1} \mathrm{x}+\tan ^{-1}(1 / \mathrm{x})$ is equal to
(a) 1
(b) $\pi / 2$
(c) $\pi / 4$
(d) 0
21. In a $\triangle A B C$, r is equal to
(a) $(S-a) \tan \frac{B}{2}$
(c) $(S-b) \tan \frac{C}{2}$
(b) $(S-b) \tan \frac{B}{2}$
22. The distance between the two lines $3 x+4 y=6$ and $6 x+8 y=15$ is
(a) $2 / 3$ units
(b) $2 / 5$ units
(c) $10 / 3$ units
(d) $3 / 10$ units
23. For what value of $\lambda$ does the equation $12 x^{2}-$

$10 x y+2 y^{2}+11 x-5 y+\lambda=0$ represents a line pair?
(a) 1
(b) 2
(c) 3
(d) 4
24. Equation of the circle having ends of its diameter at $(1,-1)$ and $(0,2)$ is
(a) $x^{2}+y^{2}-x-y-2=0$
(b) $x^{2}+y^{2}+x+y-2=0$
(c) $x^{2}+y^{2}-x-2 y+2=0$
(d) $x^{2}+y^{2}-x+2 y-2=0$
25. The direction cosines of a line equally inclined to the axes are
(a) $1 / 2,1 / 2,1 / 2$
(b) $1 / 3,1 / 3,1 / 3$
(c) $\pm 1 / \sqrt{ } 2, \pm 1 / \sqrt{ } 2, \pm 1 / \sqrt{ } 2$
(d) $\pm 1 / \sqrt{3}, \pm 1 / \sqrt{3}, \pm 1 / \sqrt{3}$

## Physics

(25 $\times 1=25$ )
26. Which of the following pair have same dimension?
(a) $L / R$ and $C R$
(b) LR and CR
(c) $\mathrm{R} / \mathrm{L}$ and $[\mathrm{LC}]^{1 / 2}$
(d) CR and $1 / \mathrm{LC}$
27. A bullet fired into a fixed target loses half of its velocity after penetrating 3 cm , the further distance travelled before coming to the rest is
(a) 4 cm .
(b) 2 cm .
(c) 3 cm .
(d) 1 cm .
28. The horizontal range is $4 \sqrt{3}$ times the maximum height in a projectile motion. The angle of projection is
(a) $15^{\circ}$.
(b) $45^{\circ}$.
(c) $30^{\circ}$.
(d) $60^{\circ}$.
29. The 'net force acting on a body is zero' then the wrong statement is
(a) body is at rest.
(b) acceleration is not zero.
(c) body is in motion.
(d) Acceleration should be zero.
30. If length of wire is doubled keeping the diameter constant, it's Young's modulus will
(a) increases.
(b) decreases.
(c) remain same
(d) depend upon nature of matter.
31. The work done to blow a soap bubble of radius ' $R$ ' is $W$, then work done to increase the radius from R to 3 R is
(a) 2 W .
(b) 8 W .
(c) 4 W .
(d) 9 W .
32. A metallic ball is immersed in alcohol. The coefficient of cubical expansion of metal is less than that of alcohol. When the system is heated weight of ball is
(a) increases.
(b) remains unchanged
(c) decreases
(d) First increases and then
decreases.
33. Latent heat of a substance is zero at
(a) boiling point.
(b) critical temperature.
(c) melting point.
(d) freezing point.
34. The average Kinetic Energy per degree of freedom per molecule of an ideal gas is
(a) $K T$.
(b) $2 K T$.
(c) $1 / 2 K T$.
(d) $3 / 4 K T$.
35. Two spheres of same material have radii in the ration 3:2. The heat radiated by them at the same temperature will be
(a) $1: 1$.
(b) $4: 9$.
(c) $9: 4$.
(d) 3:2.
36. Light of wavelength 550 nm falls normally on a slit of width $22 \times 10^{-7} \mathrm{~m}$, the angular position of second minima from central maxima will be
(a) $14.5^{0}$.
(b) $30^{0}$.
(c) $42^{0}$.
(d) $62^{0}$.
37. A person is in a room whose ceiling and two adjacent walls are mirrors. Number of images formed of an object is
(a) 5 .
(b) 7 .
(c) 6 .
(d) 8 .
38. The refractive index is 1.414 and refracting angle is $60^{\circ}$, then minimum deviation of light will be
(a) $30^{\circ}$.
(b) $60^{\circ}$.
(c) $45^{\circ}$.
(d) $72^{\circ}$.
39. A sound wave has frequency 500 Hz and velocity $360 \mathrm{~m} / \mathrm{s}$. What is the distance between 2 particles having phase difference $60^{\circ}$ ?
(a) 0.7 cm
(b) 70 cm
(c) 1.2 cm
(d) 12 cm
40. Two fixed charges $q$ and $4 q$ are at $r$ distance apart. What will be position of third charge to be placed so that the system will be in equilibrium?
(a) $2 r / 3$ from $4 q$
(b) $2 r$ from $q$
(c) $r / 2$ from $q$
(d) $r / 2$ from $4 q$
41. n-equal capacitors are first connected in series and then in parallel. The ratio of maximum to minimum capacitance is
(a) $\mathrm{n}^{2}$.
(b) $1 / \mathrm{n}^{2}$.
(c) n .
(d) $1 / n$.
42. A heater coil is cut into two equal parts and only one part is used in the heater. How will the heat generated vary?
(a) One fourth
(b) Doubled
(c) Halved
(d) Four times
43. A 50 V battery is connected across 10 Ohm resistor. The current in the circuit is 4.5 Ampere. The internal resistance of the battery should be
(a) zero
(b) 5.0 Ohm
(c) 0.5 Ohm
(d) 1.1 Ohm
44. A magnetic needle kept in a non-uniform magnetic field. It experiences
(a) a torque but not a force
(b)a force and a torque
(c) neither a force nor a torque
(d) a force but not a torque
45. In LCR circuit, the inductive reactance at resonance frequency is 100 Hz and resistance is 5 Ohm , the quality factor of the circuit is
(a) 5000 .
(b)500.
(c) 20 .
(d) 95 .
46. A circuit contains a capacitor of 420 Pf and an inductance L . The value of ' L ' to broadcast on Radio at 1020 kHz is
(a) $2.8 \times 10^{-5} \mathrm{H}$.
(b) $7.6 \times 10^{-5} \mathrm{H}$.
(c) $5.8 \times 10^{-5} \mathrm{H}$.
(d) $9.6 \times 10^{-6} \mathrm{H}$.
47. Electron accelerated from rest to a potential difference of 100 volt, its final velocity will be
(a) $5 \times 10^{5} \mathrm{~m} / \mathrm{s}$.
(b) $3 \times 10^{6} \mathrm{~m} / \mathrm{s}$.
(c) $4 \times 10^{5} \mathrm{~m} / \mathrm{s}$.
(d) $6 \times 10^{6} \mathrm{~m} / \mathrm{s}$.
48. When a proton collides with an electron, which of the following characteristics of proton increases?
(a) Energy
(b) Wavelength
(c) Frequency
(d) Impulse
49. The half life of a radioactive sample is 10 years. Its mean life is
(a) 12.43 years.
(b) 16.43 years.
(c) 14.43 years.
(d) Same as half life.
50. $N P N$ transistors are most preferred than that of $P N P$ transistor. It is because of
(a) low cost.
(b) capable of handling low power.
(c) Low dissipation of energy.
(d)high mobility of electrons than holes.

## Chemistry

( $25 \times 1=25$ )
51. Azimuthal quantum number describes
(a) orbital size
(b) orbital orientation
(c) orbital shape
(d) nuclear stability
52. The radioactive rays having highest penetrating power is
(a) $\alpha$ rays
(b) $\beta$ rays
(c) Y rays
(d) all have equal penetrating power

53. Which of the following has an electrovalent linkage?
(a) $\mathrm{MgCl}_{2}$
(b) $\mathrm{CH}_{4}$
(c) $\mathrm{SiCl}_{4}$
(d) $\mathrm{CCl}_{4}$
54. The element having highest value of electronegativity is
(a) chlorine
(b) fluorine
(c) bromine
(d) iodine
55. Oxidation number of chromium in $\mathrm{Cr}_{2} \mathrm{O}_{7}{ }^{2-}$ is
(a) +6
(b) +5
(c) +4
(d) +3
56. Which of the following is an amphoteric oxide?
(a) CaO
(b) MgO
(c) CuO
(d) ZnO
57. The structure of Buckminister fullerene contains
(a) 20 six member rings and 12 five member rings
(b) 12 six member rings and 20 five member rings
(c) 12 six member rings and 10 five member rings
(d) 12 six member rings and 8 five member rings
58. Down's process is used for the extraction of
(a) ammonia
(b) nitric acid
(c) sodium
(d) sulphuric acid
59. The IUPAC name of iso-butane is
(a) 2-methylpropane
(b) 2,2-dimethylpropane
(c) 2-methylpropene
(d)2-methylbutane
60. An alkaline $\mathrm{KMnO}_{4}$ solution is known as
(a) Tollen's reagent
(b) Baeyer's reagent
(c) Fehling's reagent
(d) Grignard reagent
61. When haloalkane solution is treated with metallic sodium in presence of dry ether, respective alkane is obtained. This reaction is known as
(a) Wurtz reaction
(b) Wolf-Kishner reduction
(c) Cannizzaro's reaction
(d) Reimer-Tiemann reaction
62. The general formula of alkanes is
(a) $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 \mathrm{n}}$
(b) $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 \mathrm{n}+2}$
(c) $\mathrm{C}_{\mathrm{n}} \mathrm{H}_{2 \mathrm{n}-2}$
(d)None of the above
63. Detection of nitrogen, sulphur and halogens in organic compounds is done by
(a) Lassaigne's test
(b) Lucas test
(c) Victor meyer test
(d) carbylamine test
64. The structure of methane is
(a) trigonal planar
(b) linear
c) tetrahedral
(d) square planar
65. Equivalent weight of potassium dichromate in an acidic medium is equal to its
(a) mol. wt./6
(b) mol. wt./3
(c)mol. wt./4
(d) mol. wt./2
66. An example of Lewis base is
(a) $\mathrm{BF}_{3}$
(b) $\mathrm{AlCl}_{3}$
(c) $\mathrm{FeCl}_{3}$
(d) $\mathrm{NH}_{3}$
67. A device which can generate electricity by means of a redox reaction is called
(a) electrochemical cell
(b) galvanic cell
(c) voltaic cell
(d)all of the above
68. 1calorie is equal to
(a) 4.184 joules
(b) 5.184 joules
(c) 3.184 joules
(d) 2.184 joules
69. The rate of a reaction depends on
(a) concentration of reacting species
(b) temperature
(c) surface area of the reactants
(d) all of the above
70. Each carbon atom of benzene is
(a) $\mathrm{sp}^{3}$ hybridized
(b) $\mathrm{sp}^{2}$ hybridized
(c) sp hybridized
(d) $\mathrm{sp}^{3} \mathrm{~d}$ hybridized
71. Which is correct for the reaction of Lucas reagent with tertiary alcohol?
(a) no turbidity occurs at room temperature
(b) turbidity appears within five minutes
(c) turbidity appears immediately
(d) turbidity appears within ten minutes

72. Fat is a
(a) lipid
(b) carbohydrate
(c) protein
(d) amino acid
73. Bakelite is made by the action of
(a) phenol on formaldehyde
(b) urea on formaldehyde
(c) melamine on formaldehyde
(d) ethylene glycol on phthalic acid
74. Which of the following compound is called corrosive sublimate?
(a) $\mathrm{Hg}_{2} \mathrm{Cl}_{2}$ (b) $\mathrm{HgCl}_{2}$
(c) $\mathrm{K}_{2}\left[\mathrm{HgI}_{4}\right]$
(d) $\mathrm{HgI}_{2}$
75. Haematite is an ore of
(a) iron(b) copper
(c) mercury
(d) silver

## English

## $(15 \times 1=15)$

76. The word "permissible" is accented on its .....syllable.
(a) $1^{\mathrm{st}}$
(b) $2^{\text {nd }}$
(c) $3^{\text {rd }}$
(d) none of the above
77. Which of the following word gets stress on the first syllable ?
(a) perish
(b) incredible
(c) incur
(d) temerity
78. I shall be glad......your help in my business.
(a) in
(b)by
(c)of
(d)with
79. An addicted person is the slave..... to his drug habit.
(a) to
(b) for
(c) on
(d) by
80. Give an appropriate synonym to the word "vendetta".
(a) pleasant
(b) secret
(c) promise
(d) feud
81. Give an appropriate synonym to the word " acerbity".
(a) mildness
(b) acrimony
(c) unsuitable
(d) extravagant
82. Give opposite meaning word to " replete".
(a) empty
(b) alarm
(c) full
(d) stuff
83. Give opposite meaning word to " quell".
(a) encourage
(b) stop
(c)crush
(d) promote
84. A disease that can be spread by touch is.......
(a) pandemic
(b)infectious
(c)endemic
(d)contagious
85. Pick the odd word out
(a)lamb
(b) pork
(c) mutton
(d) venison
86. What is the right meaning of the Latin words "ad nauseam"?
(a) to a disgusting degree (b) regularly
(c) on duty
(d) rarely
87. I think she will succeed,.....?
(a) shan't she?
(b) won't she?
(c) would she?
(d) doesn't she?
88. Which of the following is the adjective?
(a) width
(b) autumnal
(c) advantage
89. Don't stand on ceremony with him just because he has a title.

(a) avoid someone
(b) maintain formality
(c) pay too much attention to correct rules of behavior or be too formal;
(d) do every thing necessary
90. This legal document is double Dutch to me.
(a) emotional object
(b) entertainment
(c) a fair chance;
(d) difficult to understand

## Computer

$(10 \times 1=10)$
91. Analog computer works on the supply of
(a) physical strength.
(b) natural strength.
(c) magnetic strength.
(d) continuous electrical pulses.
92. We can make use of the wild card entries such as $\qquad$ in Windows Explorer.
(a) \#, @
(b) $\wedge$,*
(c) ?,*
(d) ?,\#
(c)
93. Which of the following is the second largest measurement of RAM?
(a) Megabyte
(b) Gigabyte
(c) Byte
Terabyte
(d)
94. Cache memory acts between
(a) CPU and Hard disk.
(b) RAM and ROM.
(c) CPU and RAM .
(d) All of these .
95. Which of the following is billionth of a second?
(a) Microsecond
(b) Nanosecond
(c) Terabyte
(d) Gigabyte
96. The Binary Coded Decimal (BCD) uses
(a) 6 bits
(b) 8 bits
(c) 16 bits
(d) 32 bits
97. RAM is also called as
(a) Virtual memory .
(b) Volatile memory.
(c) Non volatile memory.
(d) Cache memory.
98. CPU stands for
(a) Central Planning Unit.
(b) Computer Planning Union.
(c) Computer Processing Unit.
(d) Central Processing Unit.
99. If you want to copy a portion of the text then you need to use
(a) Control \& C
(b) Command \& C
(c) Control \& V
(d) Command and V

100 . Which one is not an operating system?
(a) Windows
(b) Linux
(c) Unix
(d) LaTex


